

REACTIONS OF ACYLGLYCINES WITH HETEROARYLHYDRAZINES

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Abstract: Reactions of *N*-acylglycines with heteroarylhydrazines leading to acylamino substituted fused 1,2,4-triazoles and *N*-{2-[2-(heteroaryl)hydrazino]-2-oxoethyl}benzamides have been investigated.

Introduction

Aminomethyl substituted heterocyclic compounds are useful compounds in heterocyclic synthesis. Their use in the synthesis of imidazo[1,5-*a*]pyrazine,^{1,3} pyrrolo[1,2-*a*]thieno[2,3-*e*][1,4]diazepine,⁴ pyrrolo[1,2-*a*][1,4]benzodiazepine,⁵ and 3a,4,5,7a-tetrahydro-5,7a-epoxyisoindoline,⁶ pyrrole,⁷ and other systems is well documented.

Recently, we reported on a new procedure for the preparation of the benzoylaminoethyl substituted fused 1,2,4-triazoles, possible precursors of the aminomethyl derivatives, based on the oxidative cyclization of *N*-(heteroarylhydrazonoethyl)benzamides.^{8,9} Since these compounds might also be prepared from hippuric acid **1** and the corresponding hydrazines, as it had been done in the case of *N*-{[(1,2,4)triazolo[4,3-*a*]pyridin-3-yl)methyl]benzamide **2**,¹⁰ we decided to reinvestigate this possibility for their preparation.

Results and Discussion

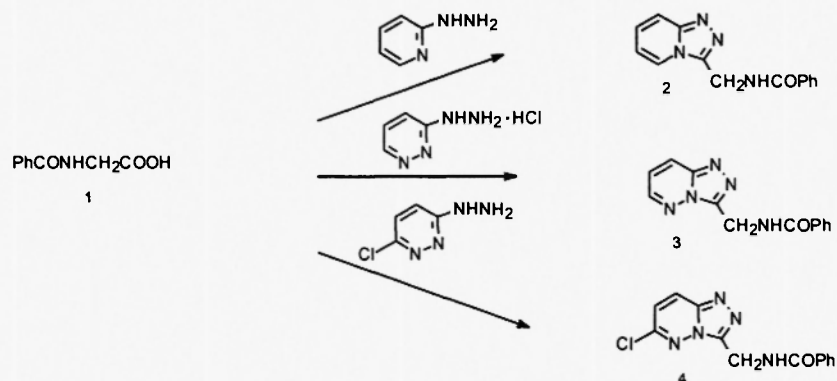
Reactions of hippuric acid **1** with heteroarylhydrazines (2-hydrazinopyridine, 3-hydrazinopyridazine hydrochloride, 3-chloro-6-hydrazinopyridazine, 2-chloro-6-hydrazinopyrazine, 6-hydrazinoimidazo[1,2-*b*]pyridazine, 6-hydrazino[1,2,4]triazolo[4,3-*b*]pyridazine, and 6-hydrazinotetrazolo[1,5-*b*]pyridazine) were carried out by two procedures.

In the procedure A, a mixture of equimolar amounts of **1** and heteroarylhydrazine was heated on an oil bath at 140-190°C under solvent-free conditions for 2-4 h. Upon cooling to room temperature the crude residue was suspended in 5% HCl and filtered. After addition of 1M NaOH to filtrate (pH=10) the separated solid was filtered off, and washed with water.

In the procedure B, a mixture of **1** (0.5 mmol) and heteroarylhydrazine (0.5 mmol) in toluene or xylene (2 ml) was heated under reflux for 8-12 h. Upon cooling to room temperature the separated solid was filtered off, washed with the same solvent and ethanol.

We found that besides 2-hydrazinopyridine, only two other hydrazines, 3-hydrazinopyridazine and 3-chloro-6-hydrazinopyridazine, gave the benzoylaminoethyl substituted fused triazoles. Isolated products **2**, **3**, and **4** were identified by comparison with authentic samples^{8,9} (Scheme 1, Table 1).

Scheme 1

Table 1. Formation of *N*-(heteroaryl(methyl)benzamides 2-4.

Product	Procedure A			Procedure B		
	Temp. [°C]	Time [h]	Yield [%]	Solvent	Time [h]	Yield [%]
2	(140-150) ^a	(4) ^a	(80) ^a	toluene	12	80
3	160-170	4	37	xylene	8	62
4	140-160	2	25 ^b	toluene	9	43 ^c

^a) lit. 10; ^b) isolated by radial chromatography (chloroform/methyl alcohol, 5:1); ^c) isolated by column chromatography (chloroform/methyl alcohol, 5:1).

Reactions with other heteroarylhydrazines under the same reaction conditions gave *N*-acylated derivatives 5-8 (Scheme 2, Table 2).

Scheme 2

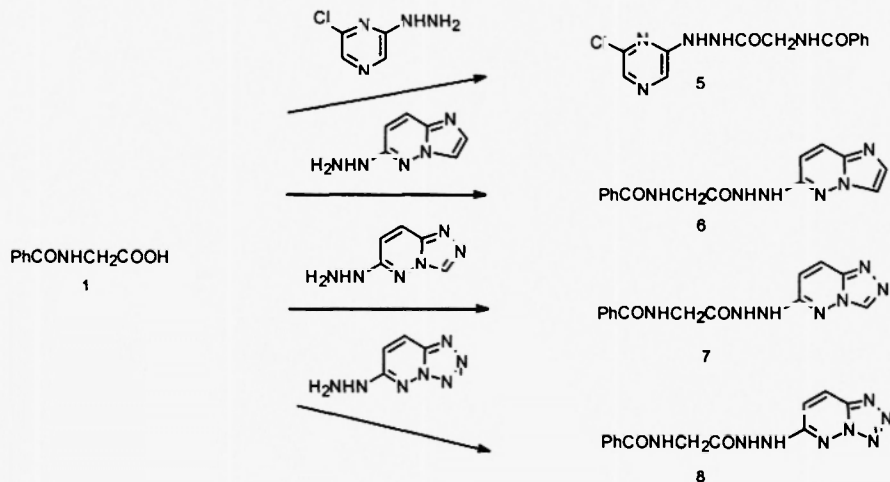
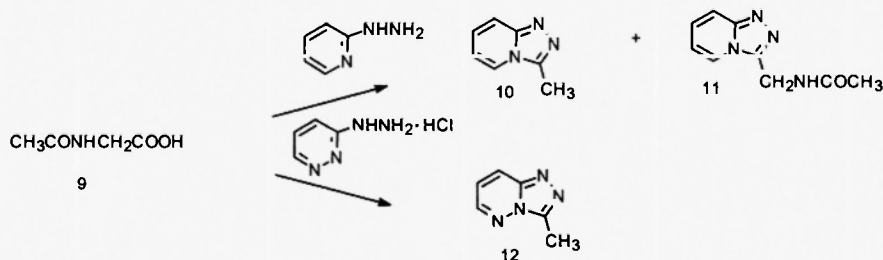


Table 2. Formation of *N*-{2-[2-(heteroaryl)hydrazino]-2-oxoethyl}benzamides 5-8.

Product	Procedure A			Procedure B		
	Temp. [°C]	Time [h]	Yield [%]	Solvent	Time [h]	Yield [%]
5 ^a	140	4	69	xylene	10.5	82
6 ^b	150-160	4	39	-	-	-
7 ^c	190	4	76	xylene	9	86
8 ^d	190	4	24	xylene	10.5	62

^a) crystallization from ethyl alcohol; ^b) radial chromatography (chloroform/methyl alcohol, 5:1); ^c) crystallization from isopropyl alcohol/methyl alcohol; d) crystallization from methyl alcohol

In two cases *N*-acetylglycine **9** was used instead of hippuric acid. Heating of **9** with 2-hydrazinopyridine at 140-150°C under solvent-free conditions for 5 h afforded methyl and acetylaminomethyl substituted [1,2,4]triazolo[4,3-*a*]pyridines, compounds **10** and **11**, isolated by extraction with chloroform and radial chromatography (chloroform) in 36 and 15% yield, respectively. Reaction of **9** with 3-hydrazinopyridazine hydrochloride under the same reaction conditions gave 3-methyl[1,2,4]triazolo[4,3-*b*]pyridazine **12**, isolated by extraction with chloroform and radial chromatography (chloroform/methyl alcohol, 10:1) in 11% yield (Scheme 3). Products **10**¹¹ and **12**¹² were identified by comparison with authentic samples prepared by known methods.

Scheme 3

Experimental

Melting points were determined on a Kofler micro hot stage and are uncorrected. NMR spectra were recorded on a Varian EM 360L (60 MHz for ¹H) or a Bruker AVANCE DPX-300 spectrometer (300 MHz for ¹H and 75.5 MHz for ¹³C) in DMSO-*d*₆ with TMS as an internal standard. Elemental analyses for C, H, N were obtained on a Perkin-Elmer CHN Analyzer 2400. MS spectra were obtained on a VG-Analytical AutoSpec Q instrument. 3-Hydrazinopyridazine hydrochloride,¹³ 3-chloro-6-hydrazinopyridazine,¹⁴ 2-chloro-6-hydrazinopyridazine,¹⁵ 6-hydrazinoimidazo[1,2-*b*]pyridazine,¹⁶ 6-hydrazino[1,2,4]triazolo[4,3-*b*]pyridazine,¹⁷ and 6-hydrazinotetrazolo[1,5-*b*]pyridazine¹⁷ were prepared as described in the literature. All other compounds were used without purification as obtained from commercial sources. *N*-{2-[2-(6-Chloropyrazin-2-yl)hydrazino]-2-oxoethyl}benzamide **5**. m.p. 200-202°C; MS (EI, *m/z*): 305 (M⁺); ¹H NMR δ 4.00 (d, 2H, *J*=5.8Hz, CH₂), 7.51 (m, 3H, Ph), 7.91 (m, 2H, Ph), 7.96 (s, 1H, H-3 or H-5), 8.03 (s, 1H, H-5 or H-3), 8.90 (t, 1H, *J*=5.8Hz, CH₂NH), 9.29 (s, 1H, NH), 10.13 (s, 1H, NH); ¹³C NMR δ 127.3, 128.2, 128.6, 131.3, 131.5, 133.8, 145.5, 155.2, 166.7, 169.0; *Anal.* Calcd. for C₁₃H₁₂ClN₃O₂: C, 51.07; H, 3.96; N, 22.91. Found: C, 50.98; H, 3.66; N, 22.62.

N-{2-[2-(Imidazo[1,2-*b*]pyridazin-6-yl)hydrazino]-2-oxoethyl}benzamide **6**. m.p. 232-234°C decomp.; MS (EI, *m/z*): 310 (M^+); $^1\text{H NMR } \delta$ 4.06 (d, 2H, $J=5.5\text{Hz}$, CH_2), 6.83 (d, 1H, $J=9.5\text{Hz}$, H-7), 7.55 (m, 4H, three H of Ph, H-2 or H-3), 7.95 (m, 4H, two H of Ph, H-3 or H-2, H-8), 8.93 (broad, 2H, two NH), 10.13 (broad, 1H, NH); *Anal.* Calcd. for $\text{C}_{15}\text{H}_{14}\text{N}_6\text{O}_2$: C, 58.06; H, 4.55; N, 27.08. Found: C, 57.71; H, 4.28; N, 26.98.

N-{2-Oxo-2-[2-([1,2,4]triazolo[4,3-*b*]pyridazin-6-yl)hydrazino]ethyl}benzamide **7**. m.p. 281-284°C; MS (EI, *m/z*): 311 (M^+); $^1\text{H NMR } \delta$ 4.03 (d, 2H, $J=5.5\text{Hz}$, CH_2), 6.92 (d, 1H, $J=9.5\text{Hz}$, H-7), 7.52 (m, 3H, Ph), 7.93 (m, 2H, Ph), 8.10 (d, 1H, $J=9.5\text{Hz}$, H-8), 8.86 (t, 1H, $J=5.5\text{Hz}$, CH_2NH), 9.18 (s, 1H, H-3), 9.37 (broad, 1H, NH), 10.13 (broad, 1H, NH); *Anal.* Calcd. for $\text{C}_{14}\text{H}_{13}\text{N}_7\text{O}_2$: C, 54.02; H, 4.21; N, 31.50. Found: C, 53.73; H, 3.99; N, 31.67.

N-{2-Oxo-2-[2-(tetrazolo[1,5-*b*]pyridazin-6-yl)hydrazino]ethyl}benzamide **8**. m.p. 231-234°C; MS (EI, *m/z*): 312 (M^+); $^1\text{H NMR } \delta$ 4.10 (d, 2H, $J=5.5\text{Hz}$, CH_2), 7.35 (d, 1H, $J=9.5\text{Hz}$, H-7), 7.57 (m, 3H, Ph), 7.99 (m, 2H, Ph), 8.51 (d, 1H, $J=9.5\text{Hz}$, H-8), 8.98 (t, 1H, $J=5.5\text{Hz}$, CH_2NH), 10.18 (broad, 2H, two NH); *Anal.* Calcd. for $\text{C}_{13}\text{H}_{12}\text{N}_8\text{O}_2$: C, 50.00; H, 3.87; N, 35.88. Found: C, 49.60; H, 3.70; N, 35.75.

N-([1,2,4]Triazolo[4,3-*a*]pyridin-3-ylmethyl)acetamide **11**. m.p. 175-178°C; MS (EI, *m/z*): 190 (M^+); $^1\text{H NMR } \delta$ 1.87 (s, 3H, CH_3), 4.28 (d, 2H, $J=5.5\text{Hz}$, CH_2), 7.02 (m, 1H, H-6), 7.39 (m, 1H, H-7), 7.80 (m, 1H, H-8), 8.48 (m, 1H, H-5), 8.65 (bt, 1H, $J=5.5\text{Hz}$, CH_2NH); *Anal.* Calcd. for $\text{C}_9\text{H}_{10}\text{N}_4\text{O}$: C, 56.83; H, 5.30; N, 29.46. Found: C, 56.57; H, 5.10; N, 29.50.

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References

1. E. Abushanab, A. P. Bindra, L. Goodman and H. Peterson, *J. Org. Chem.* **38**, 2049 (1973)
2. E. Abushanab, A. P. Bindra and L. Goodman, *J. Org. Chem.* **40**, 3379 (1975)
3. E. Abushanab, A. P. Bindra, D.-Y. Lee and L. Goodman, *J. Heterocycl. Chem.* **12**, 211 (1975)
4. A. Daich, J. Morel and B. Decroix, *J. Heterocycl. Chem.* **30**, 675 (1993)
5. D. Korakas and G. Varvounis, *Synthesis*, 164 (1994)
6. B. J. McNelis, J. T. Starr and H. Dang, *J. Heterocycl. Chem.* **35**, 1509 (1998)
7. J. J. Klappa, A. E. Rich and K. McNeill, *Org. Lett.* **4**, 435 (2002)
8. I. Music and B. Vercek, *Synth. Commun.* **31**, 1511 (2001)
9. K. Cucek and B. Vercek, *ARKIVOC* (v), 79 (2001)
10. J. B. Bicking, U. S. **3,050,525** (1962); *Chem. Abstr.* **58**, 1480e (1963)
11. J. B. Bicking, U. S. **2,917,511** (1959); *Chem. Abstr.* **54**, 8854f (1960)
12. M. Japelj, B. Stanovnik and M. Tisler, *Monatsh.* **100**, 671 (1969)
13. D. Libermann and A. Rouaix, *Bull. Soc. Chim. Fr.*, 1793 (1959); *Chem. Abstr.* **55**, 18738e (1961)
14. N. Takahayashi, *J. Pharm. Soc. Japan* **75**, 778 (1955); *Chem. Abstr.* **50**, 4970c (1956)
15. J. Bradac, Z. Furek, D. Janezic, S. Molan, I. Smerkolj, B. Stanovnik, M. Tisler and B. Vercek, *J. Org. Chem.* **42**, 4197 (1977)
16. B. Stanovnik and M. Tisler, *Tetrahedron* **23**, 387 (1967)
17. N. Takahayashi, *J. Pharm. Soc. Japan* **76**, 765 (1956); *Chem. Abstr.* **51**, 1192d (1957)

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